

N₂ fixation, CO₂ conversion, and CH₄ valorization in atmospheric pulsed plasmas

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Energy-efficient and environmentally benign alternatives for small molecules (N₂, CO₂ and CH₄) conversion are urgently needed, especially the green-chemistry, electricity-driven processes using renewable energy, to near the ultimate goal of zero carbon emissions. The paper introduces a potentially ground-breaking approach to use renewable energy in three major industrial reactions: 1) N₂ fixation, 2) CO₂ conversion and 3) CH₄ valorization, which have drastically lower carbon footprint (up to over 90% with respect to current ones). The high-energy electrons and/or gas fast-heating effect of low-temperature plasma will lower the reaction barrier and/or promote the reaction equilibrium under near ambient temperature and pressure.

1) N₂ fixation: The indirect approach involving plasma nitrogen oxidation (N₂→NO_x) in pulsed spark discharge¹ and further catalytic reduction to ammonia (NO_x→NH₃) have been proposed to approach the zero-carbon emissions target², with new insights into free-radical-chain reactions by plasma kinetics modeling and optical diagnosis (as shown in figure 1). The approach to convert N₂/H₂O directly to NH₃ in pulsed plasma-catalytic reactor is also an alternative way for Haber–Bosch process³.

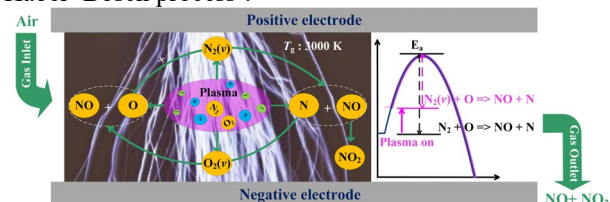


Figure 1. Sustainable plasma-driven N₂ oxidation for NO_x formation by a nanosecond pulsed spark discharge

2) CO₂ conversion: The dry reforming of CH₄ to syngas in atmospheric pulsed dielectric barrier discharge plasmas are optimized by pulse parameters (pulse rising time, width and frequency)⁴ and reactor structures (electrodes, controlled temperature and packing materials)⁵, as shown in figure 2. The methanation of CO₂ has also been implemented in atmospheric pulsed dielectric barrier discharge plasma coupling with structured catalyst and its reaction mechanism is speculated by a plasma kinetics model⁶.

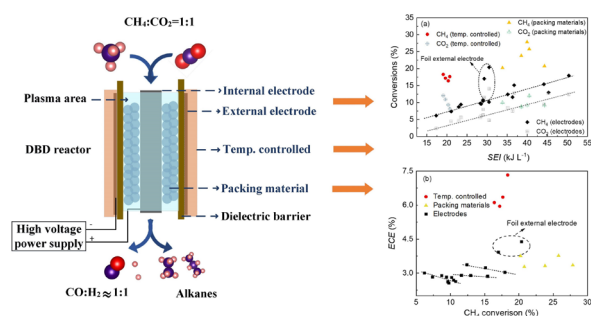


Figure 2. The dry reforming of CH₄ to syngas in atmospheric pulsed DBD plasmas

3) CH₄ valorization: We acquire a potential result that the maximum CH₄ conversion of 90% and H₂ yield of 38% could be achieved by the microsecond pulsed spark discharge with a desirable energy efficiency of 44%⁷, and the possible energy and chemical pathways are shown in figure 3. On the other hand, the dissociation CH₄ for H/CH₃ radicals by the pulsed dielectric barrier discharge plasma show the possibility for plasma-catalytic CH₄ directly to fine chemicals⁸.

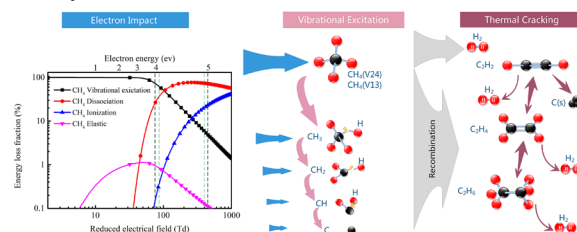


Figure 3. The schematic diagram of possible energy and chemical pathways in pulsed CH₄ spark discharge

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Note: Abstract should be in (full) double-columned one page.