

## Plasma Interactions at the Interface with Liquids, Nanoparticles and Catalytic Surfaces

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Low temperature atmospheric pressure plasmas enable the delivery of highly reactive plasma species to a substrate at (near) ambient temperatures which is beneficial for a broad range of applications. In this presentation, we will focus on advances in our understanding of (1) the underpinning mechanisms of plasma-driven solution electrochemistry enabling green nanomaterial synthesis without the need of harmful chemicals [1,2]; (2) the interaction of plasmas with catalysts as a sustainable electrically driven route to synthesize or decompose chemicals [3,5] and (3) the H<sub>2</sub> plasma-enabled reduction of iron ore as a sustainable alternative to incumbent technologies [6].

While the applications are advancing rapidly in these areas, the mechanistic understanding of these interactions is lagging mainly due to a lack of in situ diagnostic capabilities. We show that detailed experimental characterization of well-designed canonical reactors allows us to develop simplified models of complex plasma-substrate interactions leading not only to a conceptual but also quantitative understanding of the key species involved in the interactions and the rate limiting processes for plasma-liquid interactions and plasma-catalysis. Furthermore, we highlight a newly developed diagnostic capability: operando environmental Transmission Electron Microscopy (TEM) in a plasma environment to probe the mechanism of iron ore nanoparticle reduction.

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### References

- [1] P. J. Bruggeman et al, Advances in plasma-driven solution electrochemistry, *J. Chem. Phys.* (2025) 162, 071001 doi: 10.1063/5.0248579
- [2] J. H. Nam et al, Mechanisms of controlled stabilizer-free synthesis of gold nanoparticles in liquid aerosol containing plasma, *Chem. Sci.* 2024, 15, 11643, doi: 10.1039/d4sc01192a
- [3] B.N. Bayer, et al, NO formation by N<sub>2</sub>/O<sub>2</sub> plasma catalysis: A reaction-diffusion problem, *Chem. Eng. J* (2024) 482, 149041 doi: 10.1016/j.cej.2024.149041
- [4] B. N. Bayer, et al, Availability and reactivity of N<sub>2</sub>(v) for NH<sub>3</sub> synthesis by plasma catalysis, *Plasma Sources Sci. Technol.* (2023) 32, 125005 doi: 10.1088/1361-6595/ad10f0
- [5] B. Bayer, Species, pathways, and timescales for NH<sub>3</sub> formation by low temperature, atmospheric pressure plasma catalysis, *ACS Catal.* 13 (2023) 2619–2630 doi: 10.1021/acscatal.2c05492
- [6] J. Nam, Non-Thermal Plasma-Enabled Iron Oxide Reduction through Nanoscale Operando TEM (submitted)